



MOORE INSPECTIONS LLC

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## A SUMMARY OF LAB RESULTS FOR WATER SAMPLE

1234 Main St.

Buyer Name

12/09/2021 9:00AM



Inspector

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Agent

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# 1: WELL WATER TEST SUMMARY

## Information

### Understanding Your Water Sample

Understanding your water sample can be daunting. Here is a link to the Water Quality Interpretation Tool <https://erams.com/wqtool/>. Use the attached water quality laboratory analytics to fill in the tool. This tool will help you understand what is "acceptable" by EPA standards. I have also included a discussion below of a few of the more important analytes.



1731 Cherry, Helena, MT 59601  
(406)449-6292

Date Reported: 13-Oct-21

Date / Time Sampled: 07-Oct-21 @ 13:45

Date / Time Received: 08-Oct-21 @ 09:00

Chain of Custody #: 1567

Parameter	AR	NCL	SCL	Analyzed Date/Time	By	Method Reference
Total Coliform, colonies/100 mL	0	0	NR	08-Oct-21 @ 11:00	CE	SM 9223
Escol, colonies/100 mL	0	NR	NR	08-Oct-21 @ 11:00	CE	SM 9223
pH, u	8.2	NR	6.5 - 8.5	13-Oct-21 @ 13:15	CE	EPA 1501
Specific Conductivity, umhos/cm	591	NR	13-Oct-21 @ 11:45	CE	EPA 1021	
Calcium, mg/L	71.5	NR	NR	08-Oct-21 @ 11:20	CE	EPA 2007
Magnesium, mg/L	12.1	NR	NR	08-Oct-21 @ 11:20	CE	EPA 2007
Hardness, mg/L	228	NR	NR	08-Oct-21 @ 11:20	CE	Calc
Hardness, grains	13.4	NR	NR	08-Oct-21 @ 11:20	CE	Calc
Nitrate + Nitrite as N, mg/L	0.38	10	NR	11-Oct-21 @ 13:40	CE	EPA 3003
Sulfate, mg/L	107	NR	250	11-Oct-21 @ 13:40	CE	EPA 3003
Fluoride, mg/L	<0.1	0.0	2.0	11-Oct-21 @ 13:40	CE	EPA 3002
Total Dissolved Solids, mg/L	394	NR	500	13-Oct-21 @ 11:45	CE	EST 1011
Arsenic, mg/L	0.014	0.010	NR	08-Oct-21 @ 11:30	CE	EPA 2008
Cadmium, mg/L	<0.001	0.005	NR	08-Oct-21 @ 11:30	CE	EPA 2008
Chromium, mg/L	<0.001	0.100	NR	08-Oct-21 @ 11:30	CE	EPA 2008
Iron, mg/L	<0.05	NR	0.30	08-Oct-21 @ 11:30	CE	EPA 2007
Manganese, mg/L	<0.001	NR	0.050	08-Oct-21 @ 11:30	CE	EPA 2008
Lead, mg/L	0.001	0.015	NR	08-Oct-21 @ 11:30	CE	EPA 2008
Selenium, mg/L	<0.001	0.05	NR	08-Oct-21 @ 11:30	CE	EPA 2007
Uranium, mg/L	0.012	0.030	NR	08-Oct-21 @ 11:30	CE	EPA 2008

**Comments:**  
 AR - Analytical Result  
 NR - Not Regulated  
 NCL - Maximum Contaminant Limit for Drinking Water Standards  
 SCL - Secondary Contaminant Limit for Drinking Water Standards

**References:**  
 Methods for Chemical Analysis of Water and Wastewater, USEPA, 600/4-79-020  
 Method 9223 B - QT, Colibact 18, Standard Methods for the Examination of Water and Wastewater

Reviewed by: *CE*

Test Name	Lab Result	Interpretation/Acceptable	Additional Comments
Selenium (Se)	0.01 mg/L	Acceptable	This water is considered acceptable because it meets the MCL of 0.05 mg/L for selenium. Although selenium is an essential nutrient at low levels, it is toxic at high levels (such as in the case of accidental exposure). The effects of selenium in water and food depend on the amount ingested and the length of exposure. Some people who consume selenium in excess of 0.05 mg/L for extended periods of time experience hair or fingernail loss, numbness in fingers or toes, and/or coordination problems. Selenium can be treated with coagulation, reverse osmosis, ion exchange, lime softening, electrocoagulation, and/or distillation. For more information on selenium in drinking water please visit: <ul style="list-style-type: none"> <li>EPA: Basic Information about Selenium in Drinking Water</li> <li>Well Educated Fact Sheet - Cadmium and Selenium</li> <li>WHO: Chemical Fact Sheet</li> </ul>

Test Name	Lab Result	Interpretation/Acceptable	Additional Comments
Radionuclides	0.12 ug/L	Acceptable	This water is considered acceptable because it meets the MCL of 30 ug/L for uranium. Uranium is a radioactive element that will not only decay into radium and subsequently several other radioactive isotopes. During each decay alpha and beta particles are emitted. These radioactive particles can cause damage to cells and molecules in the body such as DNA. Prolonged exposure to radium may increase the risk of cancer and consumption of water containing uranium in excess of 30 ug/L may be toxic to the kidneys. Effective treatments for uranium are ion exchange, reverse osmosis, lime softening, and coagulation/filtration. For more information on uranium and other radionuclides in drinking water please visit: <ul style="list-style-type: none"> <li>EPA: Basic Information about Radionuclides in Drinking Water</li> <li>WHO: Chemical Fact Sheet</li> </ul>

\* MCL (Primary Standard)  
 \*\* MCL (Secondary Standard)  
 \*\* Upper Limit Guideline

Test Name	Lab Result	Interpretation/Acceptable	Additional Comments
1) Routine Water Analysis			
Calcium (Ca)	71.5 mg/L	No Guideline	In this state, there is no drinking water quality guideline for calcium because no documented threshold of injury has been shown to occur. Calcium is the primary constituent of water hardness. Treatment can be accomplished with a cation exchange water softener. For general information on drinking water quality please visit: <ul style="list-style-type: none"> <li>CSU Fact Sheet #113 - Domestic Water Quality - Calcium</li> </ul>
Electrical Conductivity (EC)	591 uS/cm (umhos/cm)	No Guideline	In this state, there is no drinking water quality guideline for electrical conductivity because no documented threshold of injury has been shown to occur. For general information on drinking water quality please visit: <ul style="list-style-type: none"> <li>CSU Fact Sheet #113 - Domestic Water Quality - Calcium</li> </ul>
Hardness as CaCO3	228 mg/L	Objectifiable	This water is considered objectifiable because it does not meet the upper limit guideline of 120 mg/L for hardness. Calcium and magnesium are the primary components of water hardness and are essential nutrients. High water hardness can result in scaling of bathtubs, water heaters, pipes, and other plumbing fixtures as well as soap not lathering well. The upper limit guideline of 120 mg/L is set to prevent these effects. Water hardness can be removed with an ion exchange water softener. However, most ion exchange water softeners exchange calcium and magnesium ions for sodium ions so softened water should not be used for drinking or watering plants. If softening is needed, it may be advantageous to only soften the water used for the water heater. Use a filtration system such as reverse osmosis for drinking water if all of the house's water is softened. Chemical softening, reverse osmosis, electrodeion, and ion exchange are also applicable treatments. For more information on hardness of drinking water, please visit: <ul style="list-style-type: none"> <li>CSU Extension: Drinking water quality and health</li> <li>Well Educated Fact Sheet - Hardness</li> <li>WHO: Water Softening</li> <li>WHO: Acceptability Studies</li> </ul>

Test Name	Lab Result	Interpretation/Acceptable	Additional Comments
Magnesium (Mg)	12.1 mg/L	No Guideline	At this time, there is no drinking water quality guideline for magnesium because no documented threshold of injury has been shown to occur. Magnesium is a primary constituent of water hardness. Treatment can be accomplished with a cation exchange water softener. For general information on drinking water quality please visit: <ul style="list-style-type: none"> <li>CSU Fact Sheet #113 - Domestic Water Quality - Calcium</li> </ul>
NH4-N + NO2-N (Total)	38 mg/L	Acceptable	This water is considered acceptable because it meets the MCL of 10.0 mg/L for NH4-N, NO2-N. Testing for ammonia and nitrite together as total ammonia nitrogen (TAN) is often preferred over nitrite-nitrate as N. The drinking water standard for nitrite-N is 1.0 mg/L. However, nitrite is very common in groundwater, so it is generally assumed that almost all of the nitrite plus nitrate is in the nitrate form. Natural levels of nitrate-N vary from zero to about 4 mg/L. If the value is above 4 mg/L, it is possible that nitrate-N is moving into groundwater from a surface aquifer or a septic system. Nitrate-N values exceeding 5 mg/L are a potential health concern and should be monitored regularly especially if an infant under 1 year of age is using the water. Nitrate-N values over 10 mg/L are a health concern and action should be taken to determine the source and discontinue use of the water for infants or people with cardiac problems. For more information please visit: <ul style="list-style-type: none"> <li>EPA: Basic Information about Nitrate in Drinking Water</li> <li>Well Educated Fact Sheet - Nitrate &amp; Nitrite</li> <li>WHO: Chemical Fact Sheet</li> </ul>
pH	8.2 pH	Acceptable	This water is considered acceptable because it falls within the MCL's guideline range of 6.5 to 8.5 pH. Usually does not have any direct health effects. Low pH water can be corrosive to water systems and have a bitter taste. High pH water can give water a slippery feel, a visible haze and contribute to scaling. pH can be increased by adding calcium and decreased by adding white vinegar or citric acid. Neutralizing filters can also be used (calcite or acidic ionexchange resin). For more information on pH of drinking water please visit: <ul style="list-style-type: none"> <li>Wellness - pH in Drinking Water</li> <li>Well Educated Fact Sheet - Alkalinity, pH and Total Dissolved Solids</li> <li>EPA: Secondary Drinking Water Regulations</li> <li>WHO: Chemical Fact Sheet</li> </ul>

Test Name	Lab Result	Interpretation/Acceptable	Additional Comments
Sulfate (SO4)	107 mg/L	Acceptable	This water is considered acceptable because it meets the SMCL of 250 mg/L for sulfate. Water with sulfate levels above the SMCL may have a salty taste. Sulfate concentrations greater than 1000 mg/L may have a laxative effect. Treatment can be accomplished with anion exchange, reverse osmosis, or distillation. For more information on sulfate in drinking water please visit: <ul style="list-style-type: none"> <li>CSU Extension: Drinking Water Quality and Health</li> <li>EPA: Sulfate in Drinking Water</li> <li>EPA: Secondary Drinking Water Regulations</li> <li>WHO: Chemical Fact Sheet</li> </ul>
Total Dissolved Solids (TDS)	394 mg/L	Acceptable	This water is considered acceptable because it meets the SMCL of 500 mg/L for TDS. High dissolved solids can leave deposits on pipes and fixtures, cause staining on grey water as well as taste. Treatment can be accomplished with reverse osmosis, distillation, and coagulation/filtration. For more information on TDS in drinking water please visit: <ul style="list-style-type: none"> <li>Wellness - TDS in Drinking Water</li> <li>EPA: Secondary Drinking Water Regulations</li> <li>Well Educated Fact Sheet - Alkalinity, pH and Total Dissolved Solids</li> <li>WHO: Chemical Fact Sheet</li> </ul>
2) Bacterium			
E. Coli Bacteria	0 CFU/100mL	Acceptable	This water is considered acceptable because it meets the MCL of 0 CFU/100 mL for coliform bacteria in the water. One of hundreds of strains of the bacterium Escherichia coli, E. coli O157:H7 is an emerging cause of foodborne and water-borne illness. Although most strains of E. coli are harmless and live in the intestines of healthy humans and animals, this strain produces a powerful toxin and can cause severe illness (EHEC). Treatment can be accomplished with chlorine, ultraviolet light, or ozone. For more information on E. coli in drinking water please visit: <ul style="list-style-type: none"> <li>EPA: Basic Information about E. coli O157:H7 in Drinking Water</li> <li>Well Educated Fact Sheet - Total Coliform and E. coli Bacteria</li> <li>EPA: Basic Information about Pathogens and Indicators in Drinking Water</li> </ul>

Test Name	Lab Result	Interpretation/Acceptable	Additional Comments
Total Iron	0.01 mg/L	Acceptable	This water is considered acceptable because it meets the SMCL of 0.3 mg/L for iron. Iron is an essential nutrient but at a level of 0.20 mg/L it can cause a metallic taste in the water. High iron concentrations can also cause staining of laundry and fixtures. Treatment can be accomplished with chlorine, ultraviolet light, or ozone. For more information on iron in drinking water please visit: <ul style="list-style-type: none"> <li>EPA: Basic Information about Pathogens and Indicators in Drinking Water</li> <li>Well Educated Fact Sheet - Total Coliform and E. coli Bacteria</li> <li>WHO: Water Softening</li> <li>WHO: Acceptability Studies</li> </ul>
3) Individual Metals Analysis			
Cadmium (Cd)	0.001 mg/L	Acceptable	This water is considered acceptable because it meets the MCL of 0.005 mg/L for cadmium. Long term consumption of water with cadmium at a level of 0.20 mg/L (100 mg/L in water) has been linked to kidney damage in some cases and concentrations of 1.0 mg/L may cause anemia and vomiting. There is an accepted, consistently effective method of removing cadmium at high concentrations. Lime softening may help to remove cadmium when concentrations are at or below 0.1 mg/L. For more information on this constituent in drinking water, please visit: <ul style="list-style-type: none"> <li>EPA: Basic Information about Cadmium in Drinking Water</li> <li>Well Educated Fact Sheet - Cadmium and Selenium</li> <li>WHO: Chemical Fact Sheet</li> </ul>
Chromium (Cr)	0.01 mg/L	Acceptable	This water is considered acceptable because it meets the MCL of 0.1 mg/L for chromium. Trivalent chromium may be naturally safe at a level of 0.20 mg/L. Hexavalent chromium has a deleterious effect on the liver, kidney, and respiratory organs with hematotoxic effects, dermatitis, and irritation of the skin for chronic and sub-chronic exposure. Chromium can be treated by coagulation with filtration, lime softening (Cr VI), and/or specialized processes such as ion exchange and reverse osmosis. For more information on chromium in drinking water please visit: <ul style="list-style-type: none"> <li>EPA: Basic Information about Chromium in Drinking Water</li> <li>Well Educated Fact Sheet - Chromium, Manganese, and Thallium</li> <li>WHO: Chemical Fact Sheet</li> </ul>

**Bacteria-Negative**

Water samples were found negative for coliform & E-coli. No additional actions required.

Attached are additional reports from the lab for your review and records.

**Lead-Acceptable Level**

Lead levels found within an acceptable range. The attached lab report indicates acceptable levels, with a recommended threshold of 15ppb. The link below has additional information regarding lead levels.

<https://www.epa.gov/wqs-tech/water-quality-standards-regulations-ohio>

**Nitrate/Nitrite (No2&No3) Acceptable**

The nitrogen found in the water sample is acceptable with a recommended limit of 10mg/L. No additional actions recommended at this time. Lab reports will be supplied as separate documents for your records.

**Arsenic Objectionable**

This water is considered objectionable because it does not meet the MCL of 0.01 mg/L. Arsenic is poisonous in humans at 100 mg or more and has proven lethal at 130 mg. Studies have linked long-term exposure to arsenic in drinking water to cancer of the bladder, lungs, skin, kidney, nasal passages, liver, and prostate. Non-carcinogenic effects of arsenic include cardiovascular, pulmonary, immunological, neurological, and endocrine effects. Short-term exposure to high doses of arsenic can cause other detrimental health effects, but are unlikely to occur from public water supplies in compliance with the existing arsenic standard of 0.01 mg/L. Accumulation in the body is expected to rise progressively in humans with low intake of arsenic. By using activated alumina, reverse osmosis, ion exchange, or electro dialysis, the concentration of arsenic can be significantly lowered. For more information about arsenic in drinking water please visit:

- [EPA: Basic Information about Arsenic in Drinking Water](#)
- [Well Educated Fact Sheet- Arsenic](#)
- [Butte-Silver Bow Water Environmental Health](#)

## 2: SAMPLING PROTOCOL

### Information

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#### How I Sampled Your Well

I follow the US EPA Quick Guide to Drinking Water Sample Collection. Follow the link here:  
[https://www.epa.gov/sites/default/files/2015-11/documents/drinking\\_water\\_sample\\_collection.pdf](https://www.epa.gov/sites/default/files/2015-11/documents/drinking_water_sample_collection.pdf)